

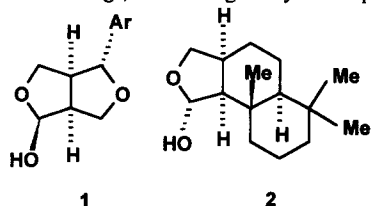
Intramolecular [2+2] Photocycloaddition for the Direct Stereoselective Synthesis of Cyclobutane Fused γ -Lactols

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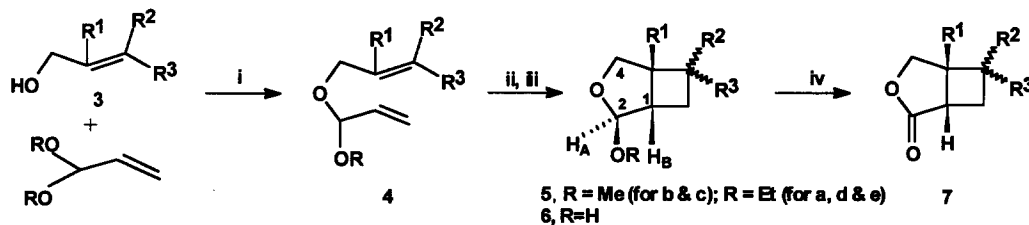
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Abstract : A direct stereoselective route for the synthesis of γ -lactols fused to cyclobutanes via intramolecular [2+2] photocycloaddition of 1,6-dienes is described.
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γ -Lactols fused to other rings are frequently encountered in a wide range of natural products. Representative examples include the lignan saminin **1**¹ and the sesquiterpene drimane **2**.² In addition, γ -lactols serve as valuable reactive intermediates in organic synthesis.³ The usual way of annulating lactols involves either intramolecular hemiacetalisation⁴ of appropriately disposed aldehyde and alcohol functionalities in a prebuilt ring, or reduction of lactones⁵ and proceeds nonstereoselectively in many occasions. We here report a topologically different strategy that results in the direct stereoselective synthesis of cyclobutane fused γ -lactols which, by virtue of the properties of two reactive rings, will be of great synthetic potential.



The key step in the present strategy is the cycloaddition of two π units linked together through an acetal oxygen. The overall synthetic plan is illustrated with the synthesis of the lactol **6a** (Scheme 1). Acid catalysed trans-acetalisation of acrolein diethyl acetal with the allyl alcohol **3a** in refluxing benzene with azeotropic removal of ethanol afforded the mixed acetal **4a**⁶ in 68% yield. Cycloaddition⁷ of the diene **4a** could be achieved through irradiation of its ether solution in the presence of cuprous trifluoromethane sulfonate (CuOTf) as the catalyst to afford the protected lactol **5a** in 50% yield thus demonstrating the compatibility of the acetal functionality to CuOTf catalysis in the photocycloaddition. The *exo*-stereochemical assignment to the photoadduct **5a** was based on comparison of the coupling constant of H_A ($J=0$) to those reported⁸ for analogous *exo* ($J_{A,B-trans} = 1.5$ Hz)- and *endo* ($J_{A,B-syn} = 5.5$ Hz)-2-substituted-3-oxa-bicyclo[3.2.0]heptanes. Brief exposure of the adduct **5a** to hot 80% aqueous acetic acid led to smooth deprotection to afford the lactol **6a** in 71% yield. Retention of configuration during deprotection was indicated by the appearance of a singlet at δ 5.19 and is attributed to be the result of addition of H₂O to the intermediate oxonium ion **8a** from the least hindered *exo* face. The generality of this three step protocol was established by the synthesis of a series of lactols **6b-e** in very good yields (Table 1). It is to be noted that deprotection



a, R¹=H, R²=R³=Me; b, R¹=R²=R³=H; c, R¹=Me, R²=R³=H; d, R¹=H, R²,R³=H, Me; e, R¹=H, R²,R³=H, CH=CHMe

Scheme 1. Reagents and conditions : i) PPTS, C₆H₆, Δ . ii) hv, CuOTf, Et₂O, iii) HOAc-H₂O, Δ , iv) Jones reagent.

of the lactol **5c** (entry3) resulted in inversion of configuration at C₂ to lead to the thermodynamically more stable lactol **9** ($J_{A,B} = 9.9$ Hz), arising through addition of H₂O to the oxonium ion **8c** from the *endo* face as addition from the *exo* face is blocked by the Me group. The lactols **5** or **6**, on Jones oxidation, could be transformed to the cyclobutane fused lactones **7**⁹ which are also useful synthetic intermediates.

To demonstrate the synthetic potential of the cyclobutane fused lactols, the lactone **7c**, obtained as above, was transformed to the diol **10** (71%) on reaction with excess MeLi in ether. Swern oxidation of the diol **10** afforded a diastereoisomeric mixture of the lactols **11** (63%) which has already been transformed³ to grandisol **12**,¹⁰ a component of the aggregation pheromone of the Boll weevil.

In conclusion, trans-acetalisation of an acrolein acetal with allyl alcohols in conjunction with photocycloaddition offers an excellent stereoselective route to cyclobutane fused lactols.

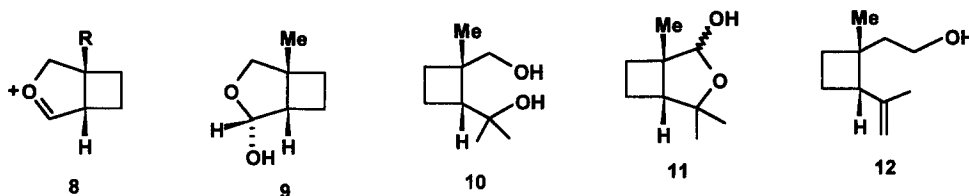


Table 1. Synthesis of cyclobutane fused lactols and lactones

Entry	Dienes	Lactols	Lactones
1	4a (68) ^a	5a (50), 6a (71)	7a (68)
2	4b (72)	6b (59) ^b	7b (60)
3	4c (68)	9 (56) ^b	7c (70)
4	4d (74)	5d (69) ^c , 6d (77) ^c	7d (72)
5	4e (77)	5e (70) ^d , 6e (70) ^d	7e (71)

^a Figures in the parenthesis represents yields of products isolated by chromatography. ^b The photoadducts were directly hydrolysed without characterisation. ^c Obtained as a mixture of two diastereoisomers in 2.5:1 ratio with *exo*-Me isomer as the major one.

^d Obtained as a mixture of four components in 5:2.5:1.5:1 ratio arising from two diastereoisomers of each of the two geometrical isomers.

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References and Notes

1. Ward, R.S. *Nat. Prod. Rep.* **1999**, *16*, 75.
2. Pulici, M.; Sugawara, F.; Koshino, H.; Usawa, J.; Yoshida, S.; Lobkovsky, E.; Clardy, J. *J. Nat. Prod.* **1996**, *59*, 47.
3. Hoffmann, N.; Scharf, H.D. *Liebigs Ann. Chem.* **1991**, 1273.
4. Binns, F.; Hayes, R.; Ingham, S.; Saengchantara, S.T.; Turner, R.W.; Wallace, T.W. *Tetrahedron* **1992**, *48*, 515.
5. Samajdar, S.; Patra, D.; Ghosh, S. *Tetrahedron* **1998**, *54*, 1789.
6. All new compounds were characterised through ¹H (300 MHz) and/or ¹³C NMR (75 MHz) and IR (in appropriate cases) spectroscopy. NMR data for selected compounds : **5a**, ¹H δ 0.93 (s, 3H), 1.13 (s, 3H), 1.16 (t, $J=7.2$ Hz, 3H), 1.73 (m, 2H), 2.45 (m, 1H), 2.77 (m, 1H), 3.41 (q, $J=6.9$ Hz, 2H), 3.74 (dd, $J=6.3$ and 9.6 Hz, 1H), 3.96 (d, $J=9.6$ Hz, 1H), 4.82 (s, 1H); ¹³C 15.3, 23.9, 31.3, 32.6, 35.5, 38.5, 46.6, 62.1, 67.1, 107.4. **6a**, ¹H δ 0.87 (s, 3H), 1.09 (s, 3H), 1.67-1.74 (m, 2H), 2.45 (t, $J=6.6$ Hz, C₅-H), 2.71-2.79 (m, 1H), 3.87 (dd, $J=6.6$ and 9.6 Hz, C₄-H), 3.96 (d, $J=9.6$ Hz, C₄-H), 5.19 (s, 1H).
7. For our previous work on the CuOTf catalysed photocycloaddition see : a) Ghosh, S.; Patra, D.; Saha, G. *J. Chem. Soc. Chem. Commun.* **1993**, 783. b) Patra, D.; Ghosh, S. *J. Org. Chem.* **1995**, *60*, 2526. c) Ghosh, S.; Patra, D.; Samajdar, S. *Tetrahedron Lett.* **1996**, *37*, 2073. d) Haque, A.; Ghatak, A.; Ghosh, S.; Ghoshal, N. *J. Org. Chem.* **1997**, *62*, 5211. e) Samajdar, S.; Ghatak, A.; Ghosh, S. *Tetrahedron Lett.* **1999**, *40*, 4401.
8. Gregori, A.; Alibes, R.; Bourdelande, J.L.; Font, J. *Tetrahedron Lett.* **1998**, *39*, 6961.
9. Ghosh, S.; Raychaudhuri, S.R.; Salomon, R.G. *J. Org. Chem.* **1987**, *52*, 82 and the references cited therein.
10. For some recent syntheses see: a) Langer, K.; Mattay, J. *J. Org. Chem.* **1995**, *60*, 7256. b) Monteiro, H.J.; Zukerman-Schpector, J. *Tetrahedron* **1996**, *52*, 3879. c) Alibes, R.; Bourdelande, J.L.; Font, J.; Parella, J. *Tetrahedron* **1996**, *52*, 1279.